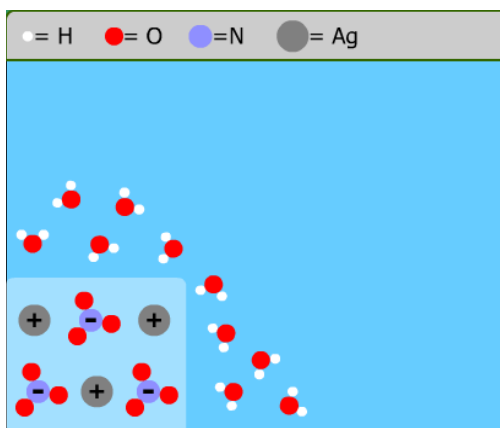
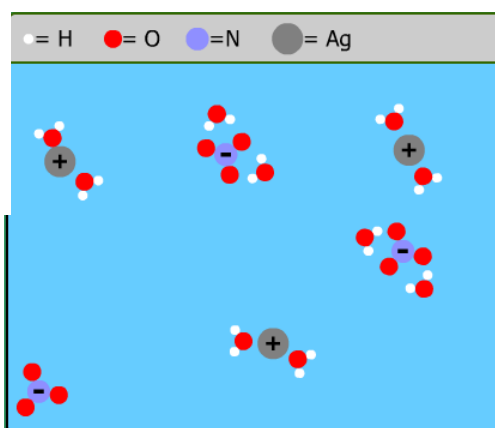


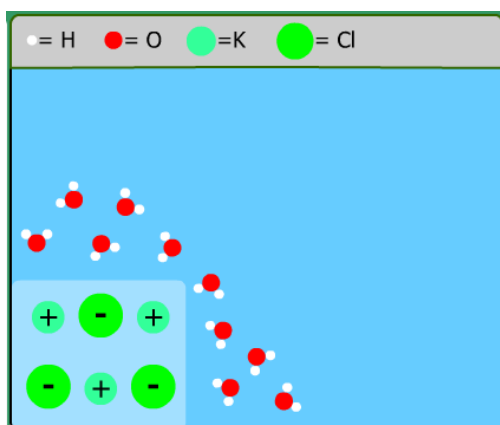
ChemThink: Chemical Reactions Precipitate Lab



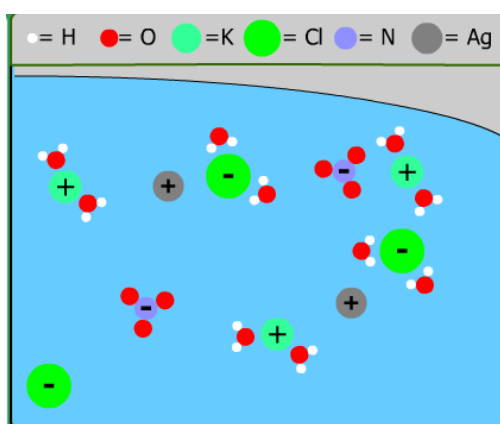
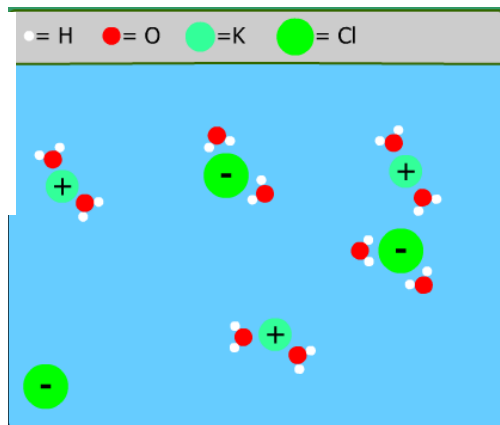
Describe what is happening in the two pictures.



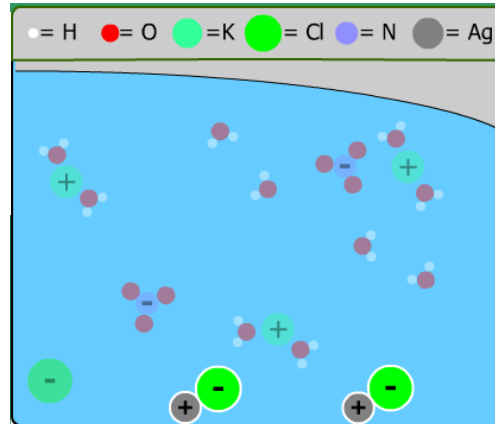
If you have a solution marked $\text{AgNO}_3(\text{aq})$, what is really in the bottle?



Describe what is happening in the two pictures.



Describe what is happening in the two pictures.



Why do some solutions form precipitates (pieces of solids) when they are mixed?

What does a complete ionic equation show? Give an example.

What does a net ionic equation show? What is left out of the equation?
Give an example.

What does a molecular equation show? Give an example.

Why do some solutions do nothing (don't form pieces of solid) when they are mixed?